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ion ( $\text{H}_3\text{O}^+$ ). The autoionization of liquid water produces  $\text{OH}^-$  and  $\text{H}_3\text{O}^+$  ions.

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acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightleftharpoons \text{H}_2\text{O} + \text{NO}_3^-$ .  $\text{HNO}_3$  and  $\text{NO}_3^-$  make one pair  $\text{OH}^-$  and  $\text{H}_2\text{O}$  make the other.

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